

INVESTIGATIONS INTO THE ICE NUCLEATING ABILITY OF PROPANE FLAME SOOT

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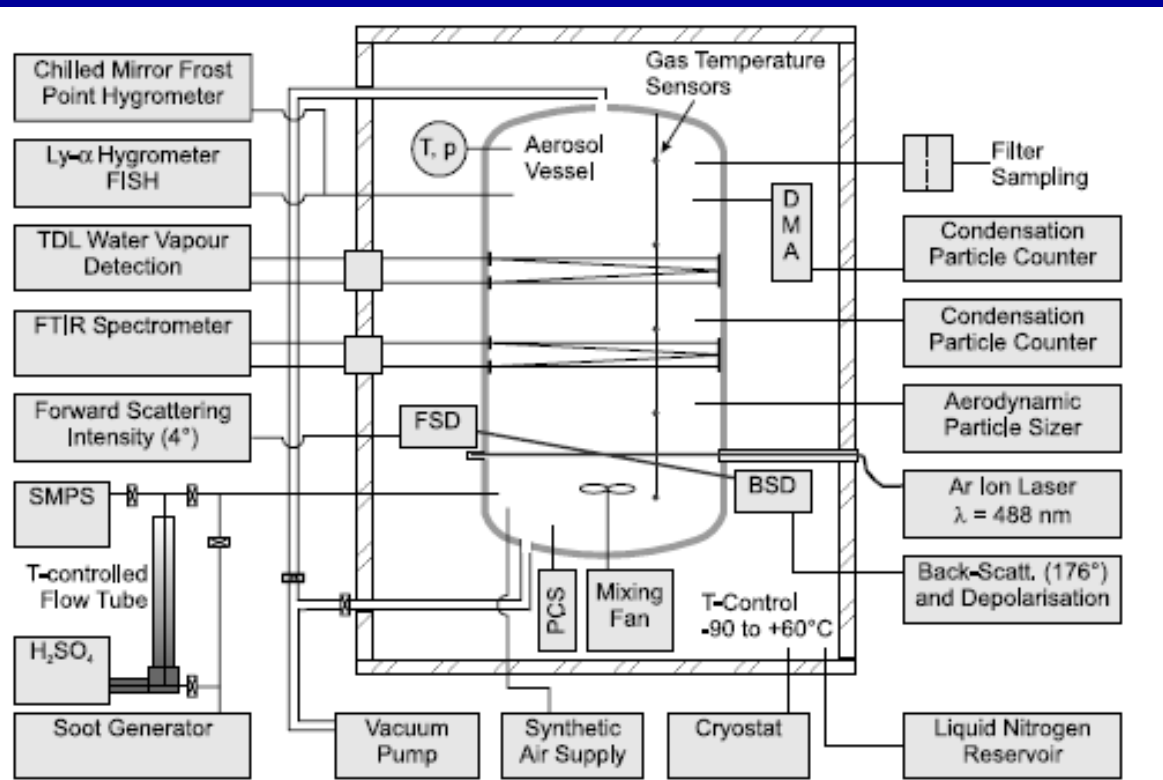
VI-ACI Workshop 2008

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Outline of Talk

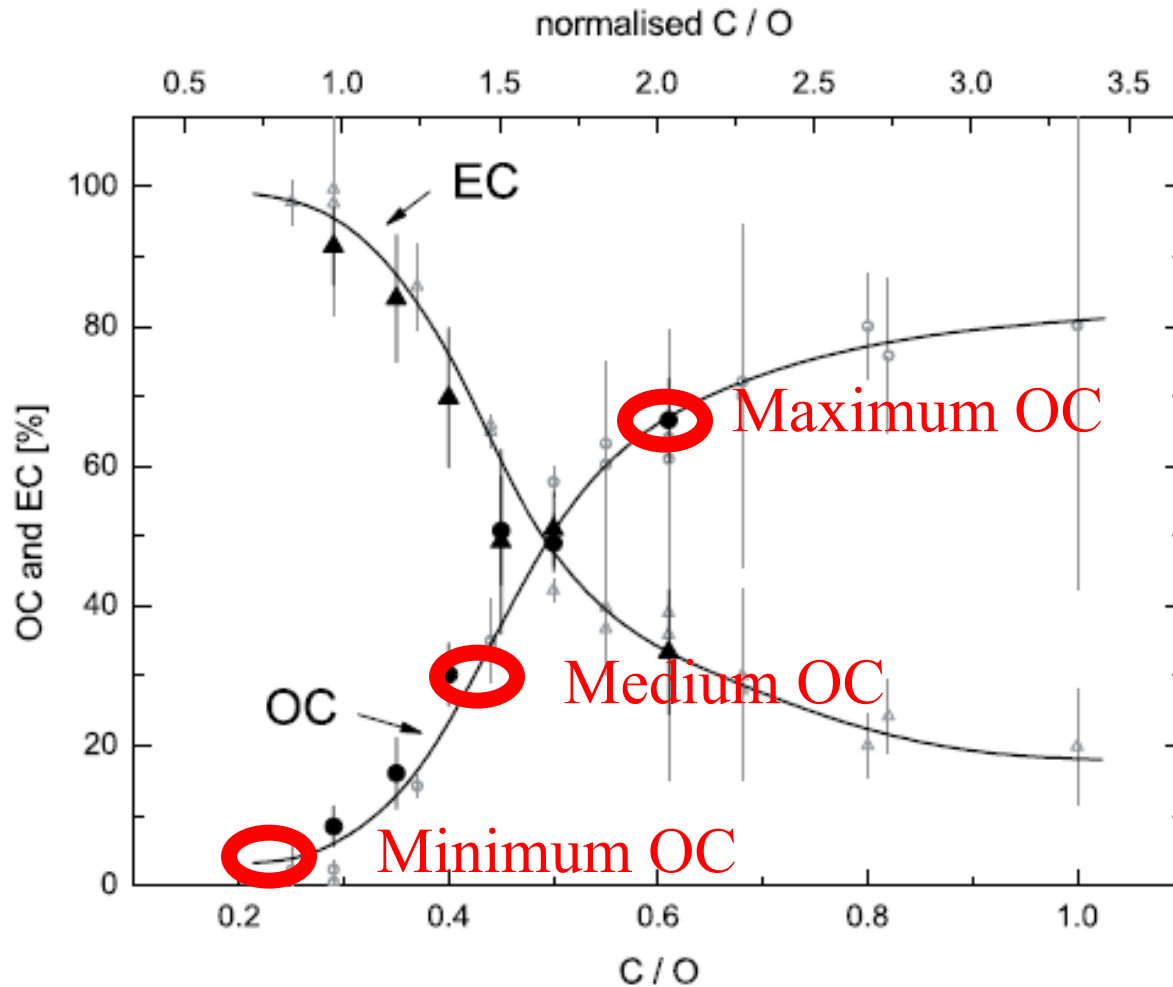
- AIDA Chamber
- Instrumentation
- Experiments
- Results
- Conclusions

AIDA Chamber



- Volume 84.3m^3
- Operational temperature range $183 \leq T \leq 333 \text{ K} \pm 0.5 \text{ K}$
- Walls coated with ice
- Semi-adiabatic expansion reduced chamber temperature \rightarrow increases relative humidity.

CAST Soot Generator



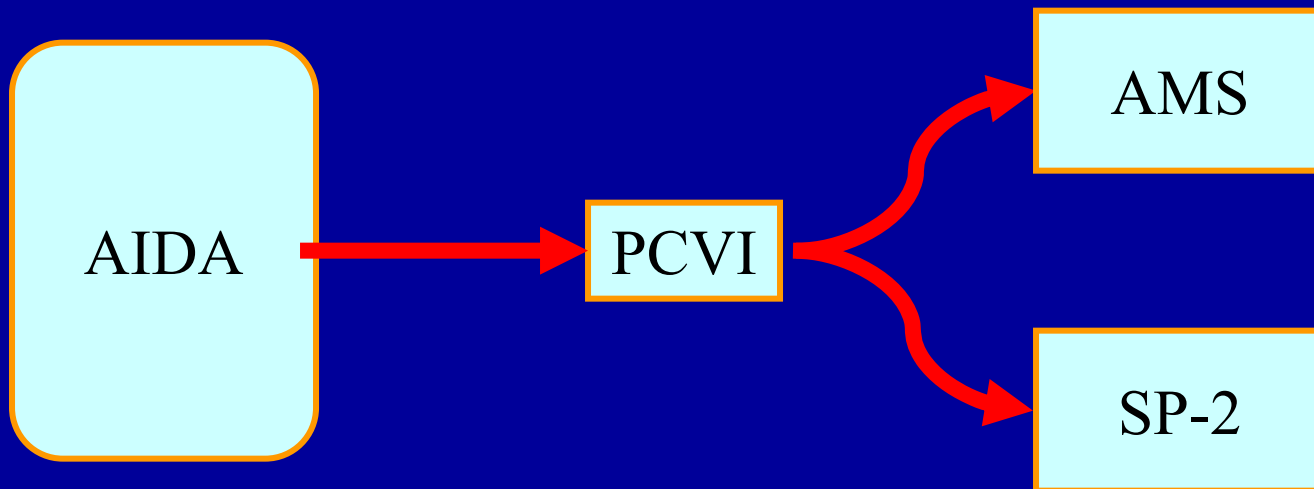
- Propane Flame soot generator. Different OC achieved by altering fuel:air mixture

- Minimum ~ 5% OC

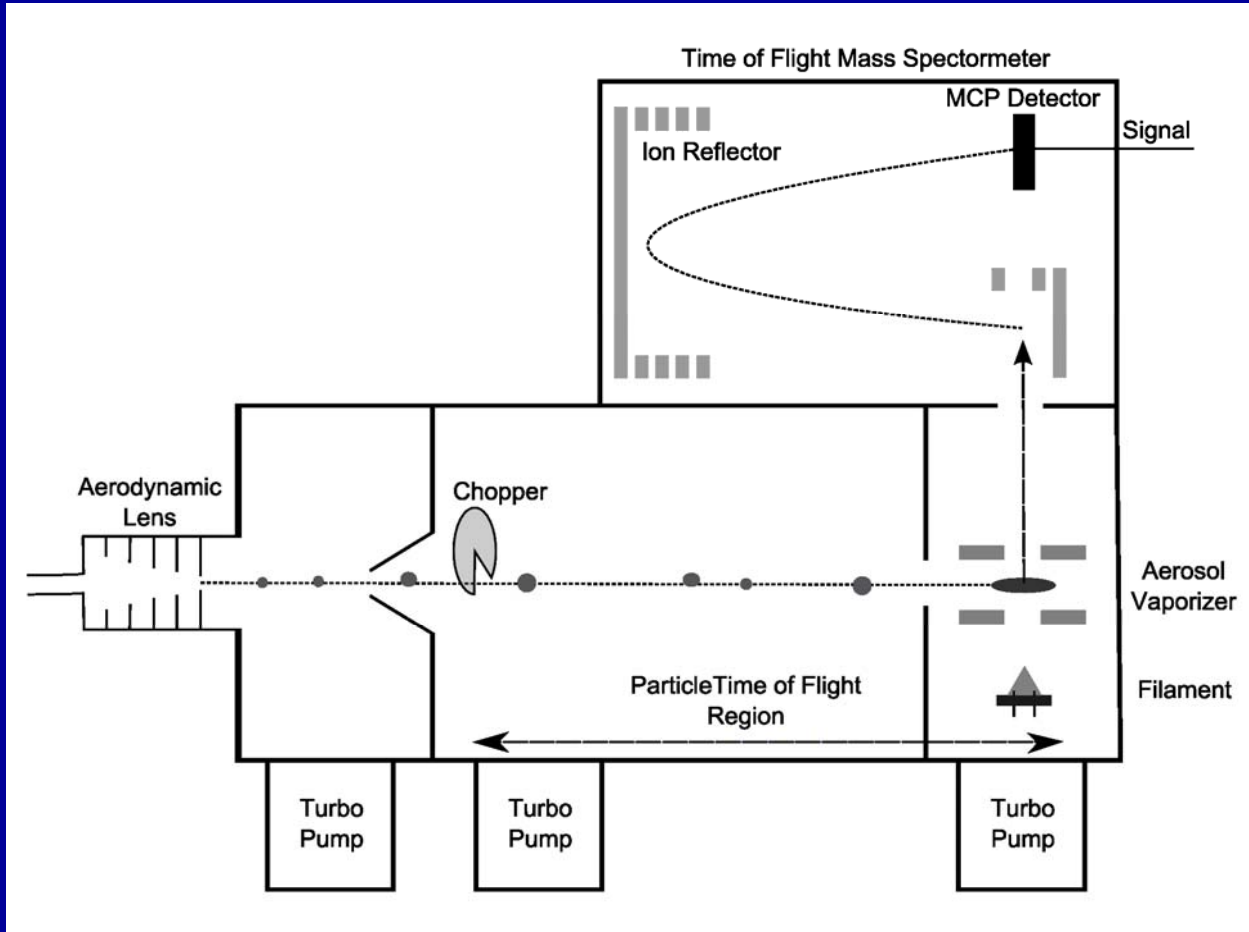
- Medium ~ 30% OC

- Maximum ~ 60-70% OC

Instrumentation

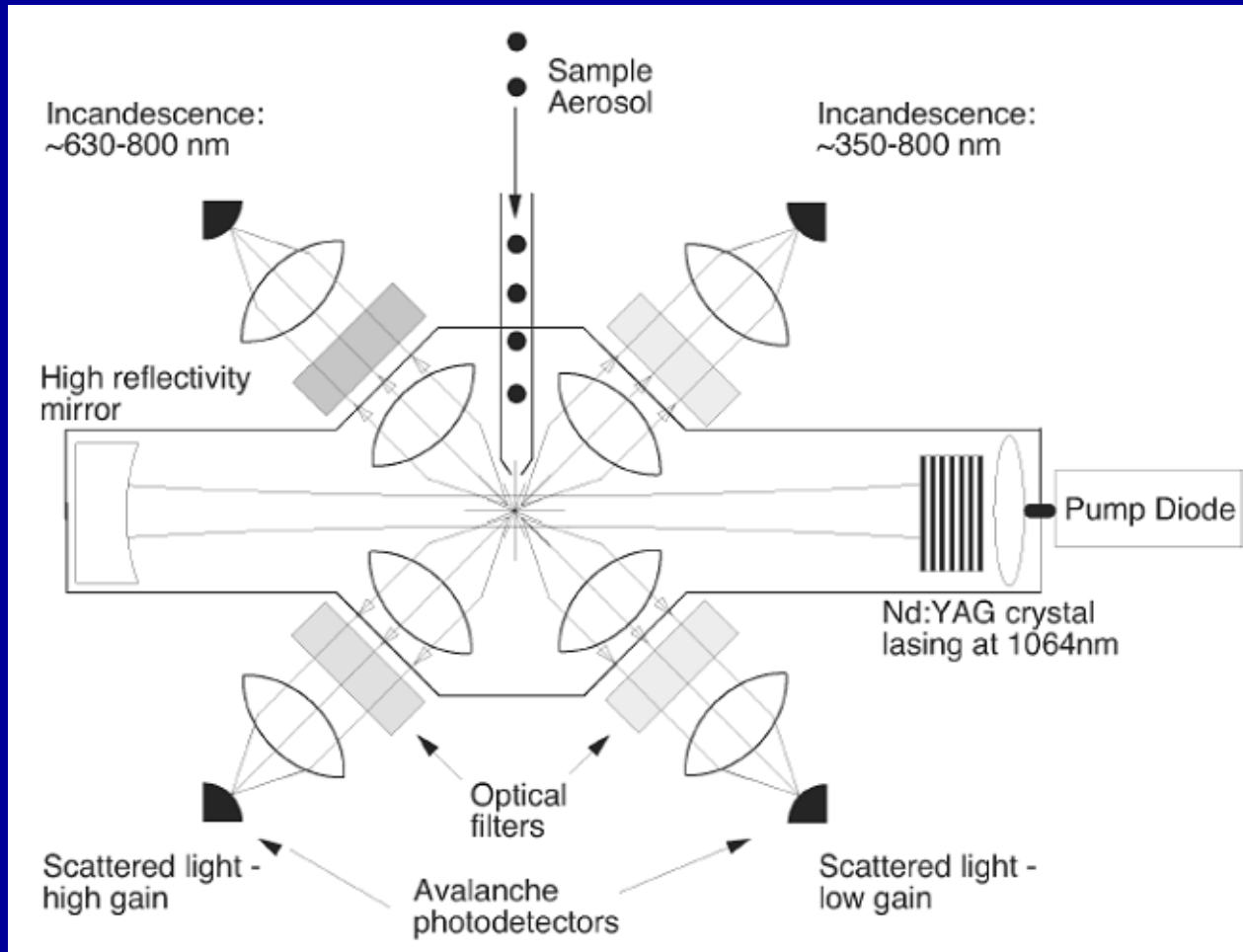


Time of Flight AMS

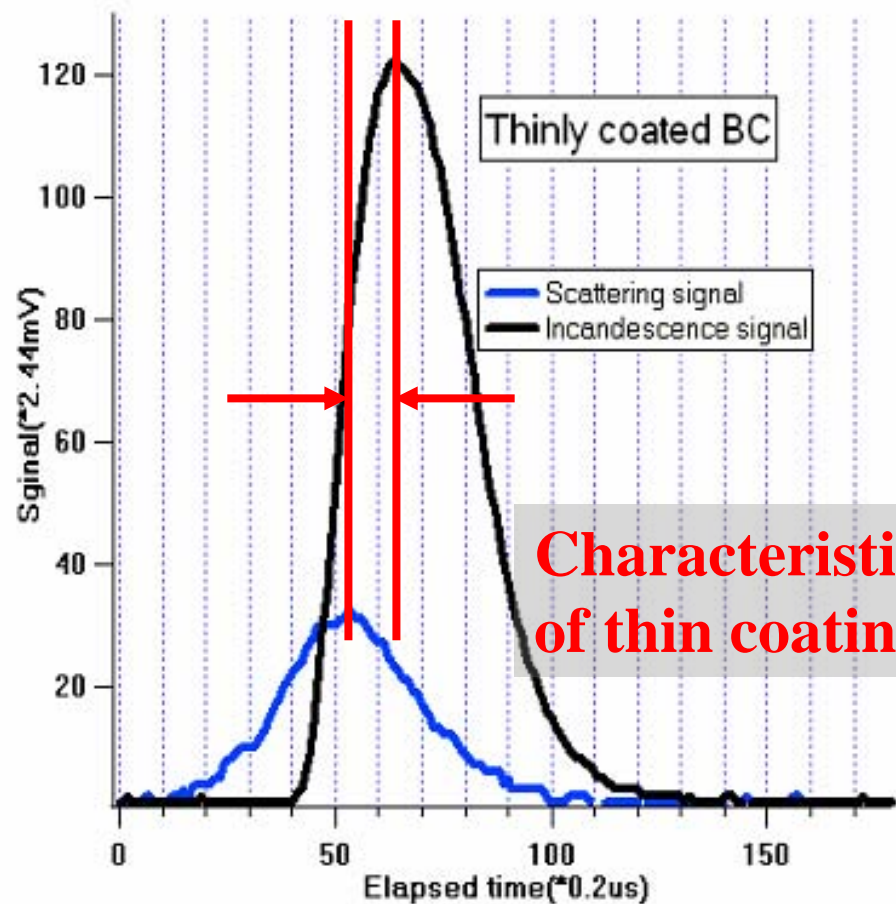


determines the size and composition of aerosol

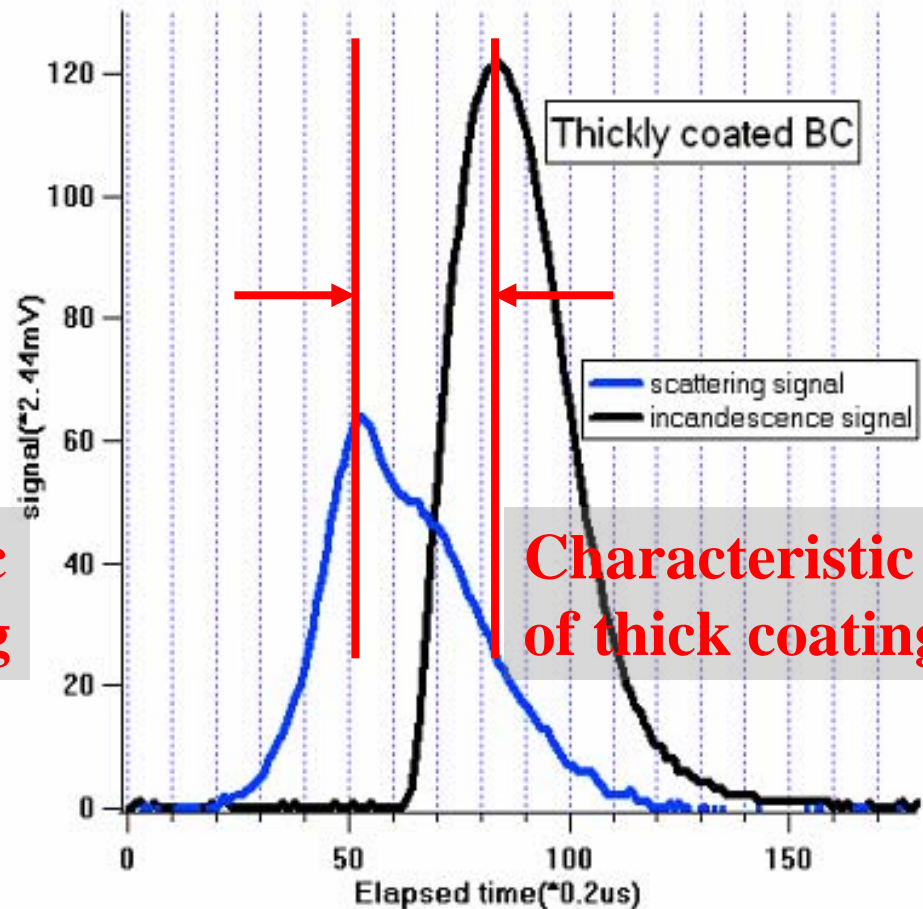
Single Particle Soot Photometer



determines the size, mass and incandescing temperature of aerosol particles $0.15-1\mu\text{m}$ in diameter on a single particle basis

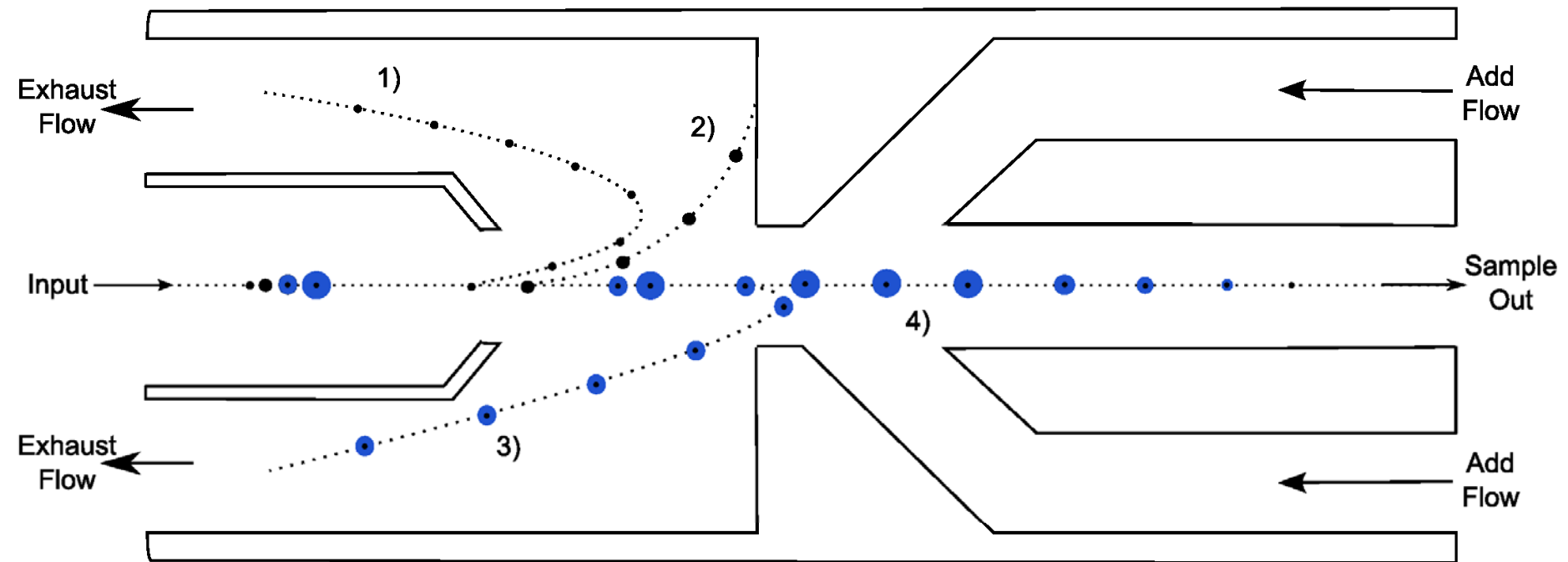


**Characteristic
of thin coating**



**Characteristic
of thick coating**

Pumped Counterflow Virtual Impactor



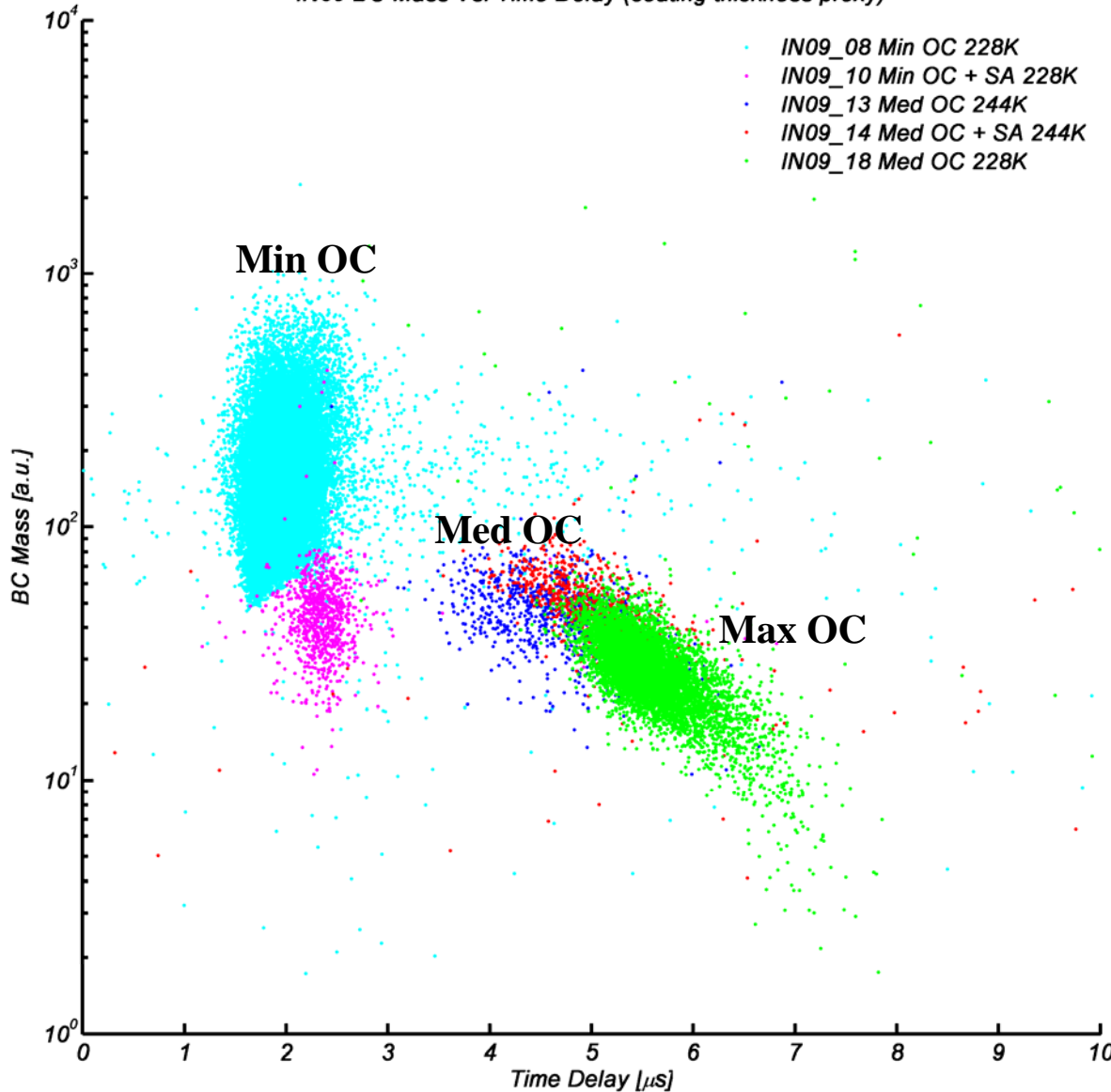
- 1) Small aerosol follows the exhaust flow stream lines - Not Sampled
- 2) Larger aerosol are knocked off a sampling trajectory by the counterflow and are impacted - Not Sampled
- 3) Small hydrometeors and large aerosol do not have sufficient inertia to pass stagnation region - Not Sampled
- 4) Sufficiently large hydrometeors pass stagnation region and evaporate - Sampled

Experiments

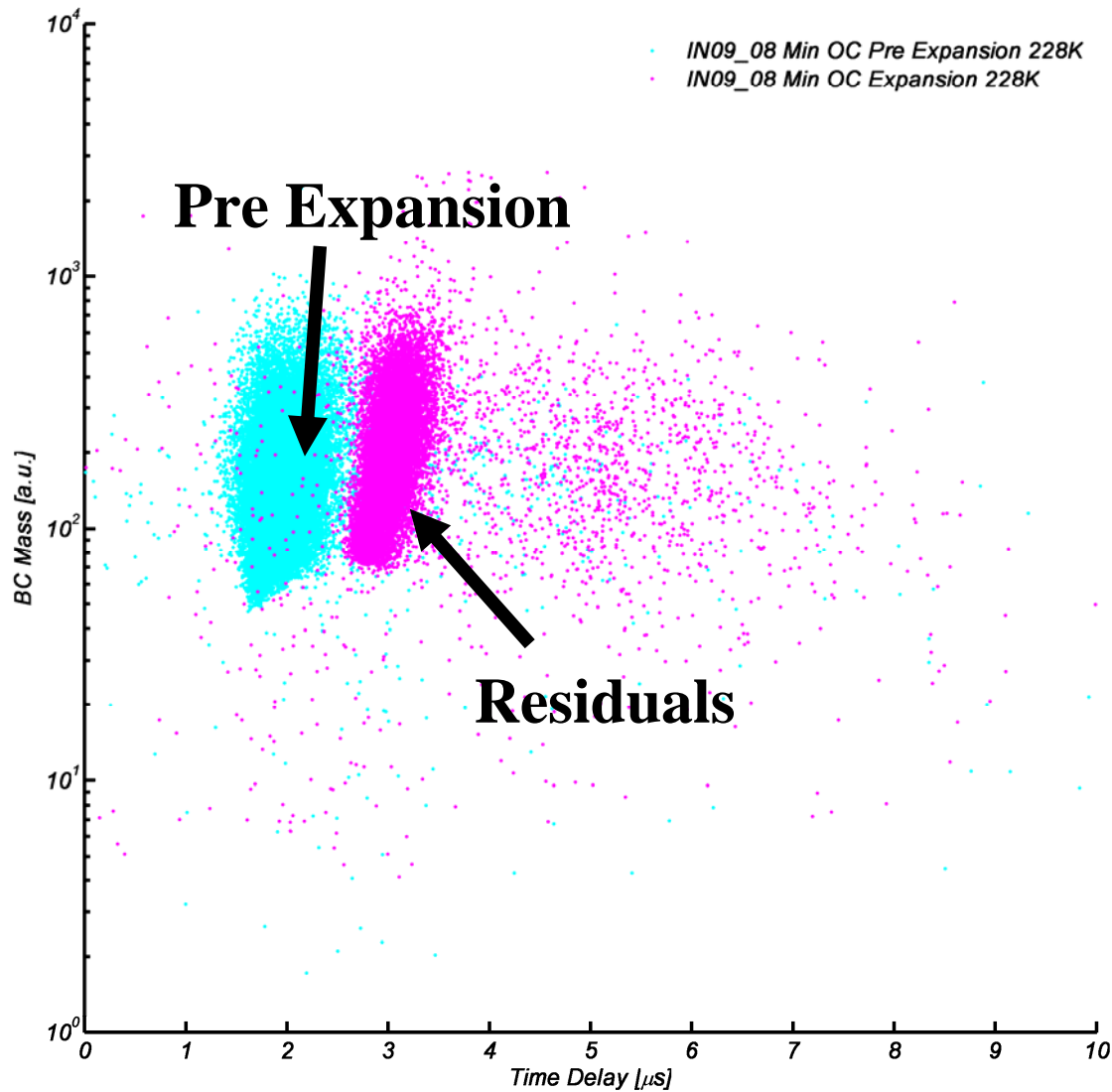
1. Influence of Soot Organic Carbon Content on Ice Nucleation Efficiency
228K, Pumping Speed Varied
2. Influence of Externally Mixing Sulphuric Acid with Soot on Ice Nucleation Efficiency
228K & 244K

Pre expansion Coating thickness and OC

IN09 BC Mass Vs. Time Delay (coating thickness proxy)



- Delay between scattering and incandescing signals gives a proxy for the coating thickness of the soot.
- Minimum OC soot has thinnest coating.
- Coating thickness increases with OC.
- Expect Min OC to be of a *fractal* morphology.
- Med and Max OC are likely to be non-fractal *blobs*.



- Significant difference in coating thickness proxy between pre expansion and residual soot.

- Either the soot is gaining coating during nucleation – unlikely

OR

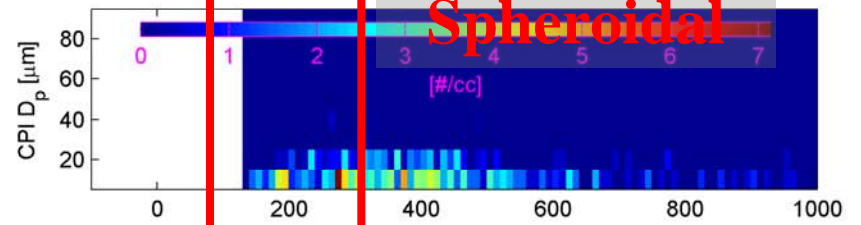
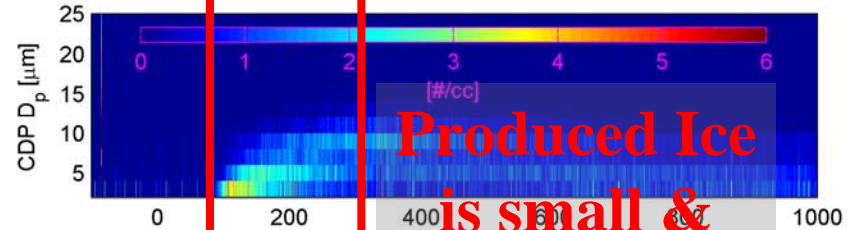
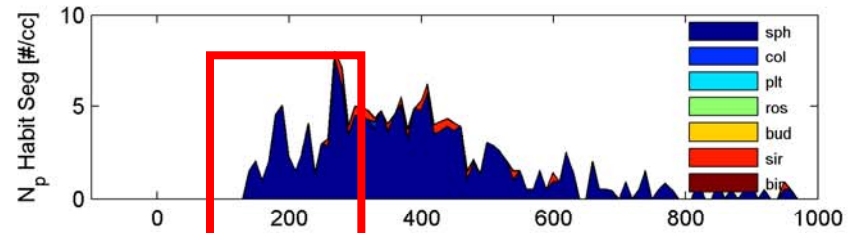
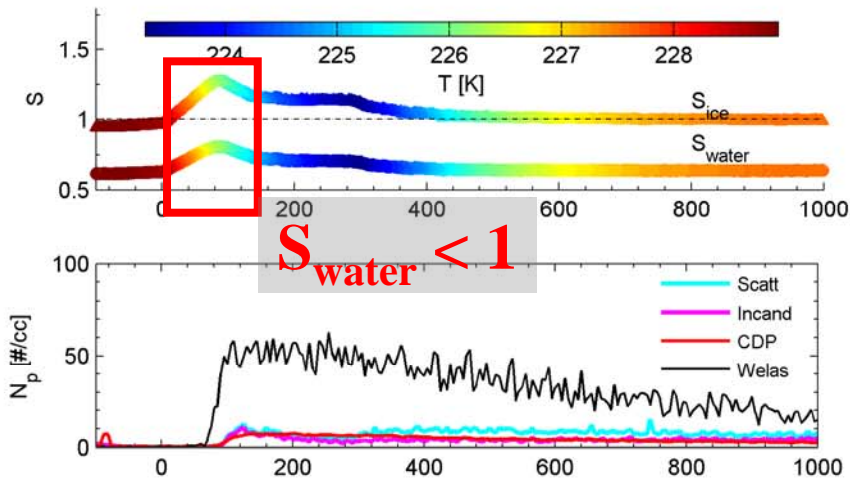
- Volatiles are condensing on the soot - possible

OR

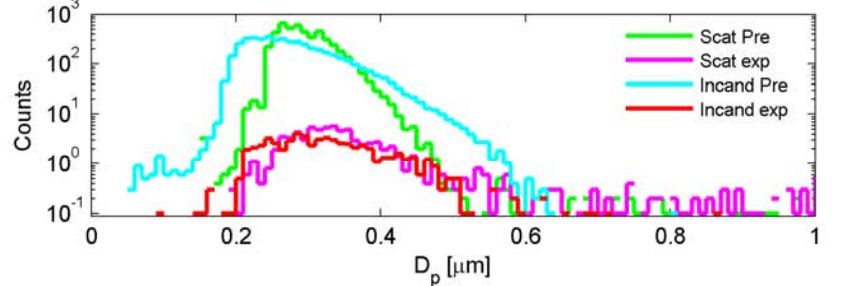
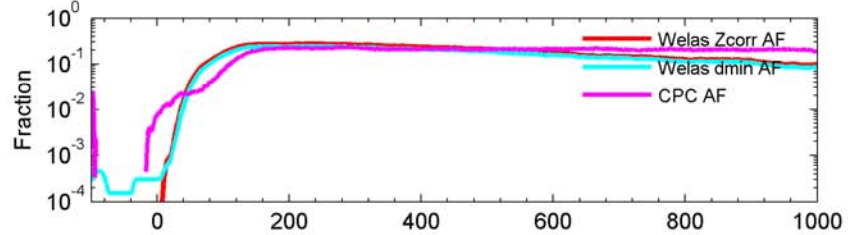
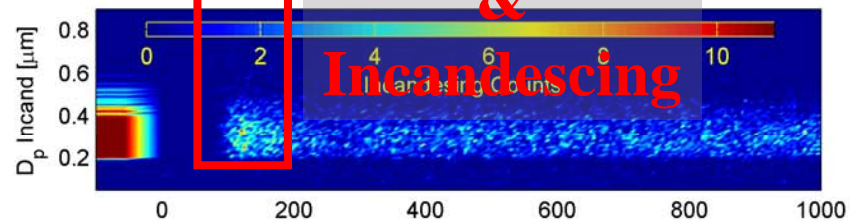
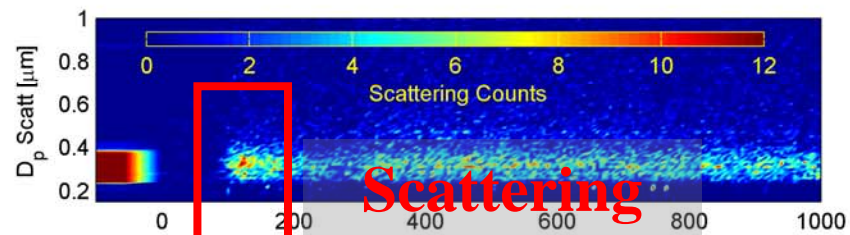
- Change in morphology is occurring (fractal to blob) which is changing the incandescing time - probable

Influence of coating thickness

228K



30-35% Activation



0,00-10,00

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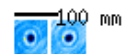
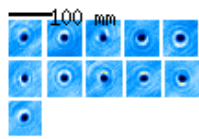
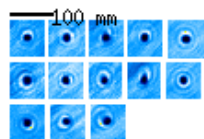
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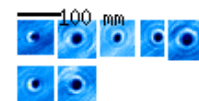
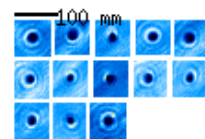
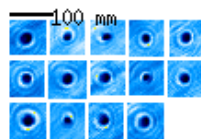
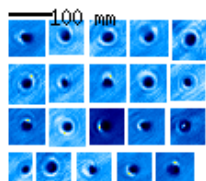
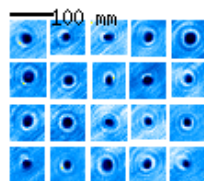
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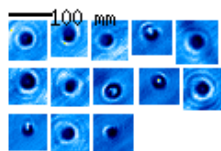
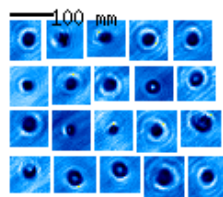
10,00-20,00



20,00-30,00

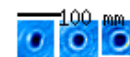


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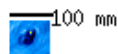
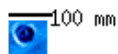
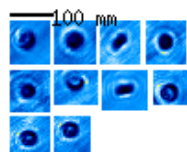


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No Data

No Data

No Data

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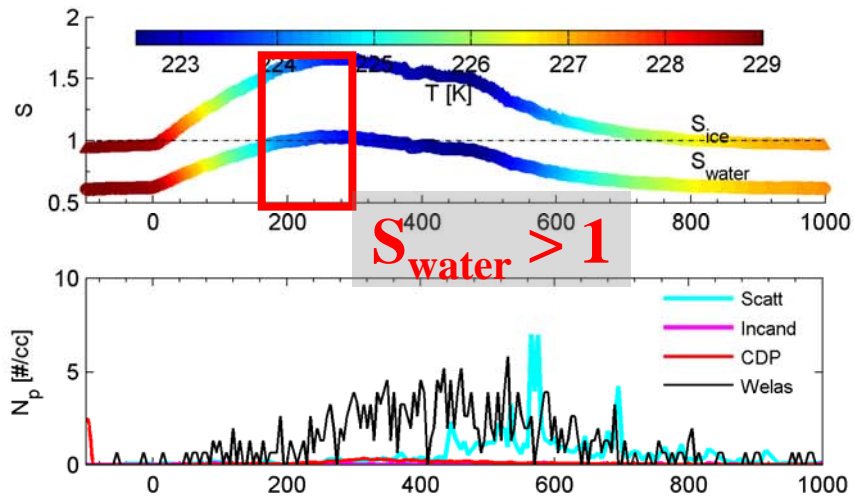
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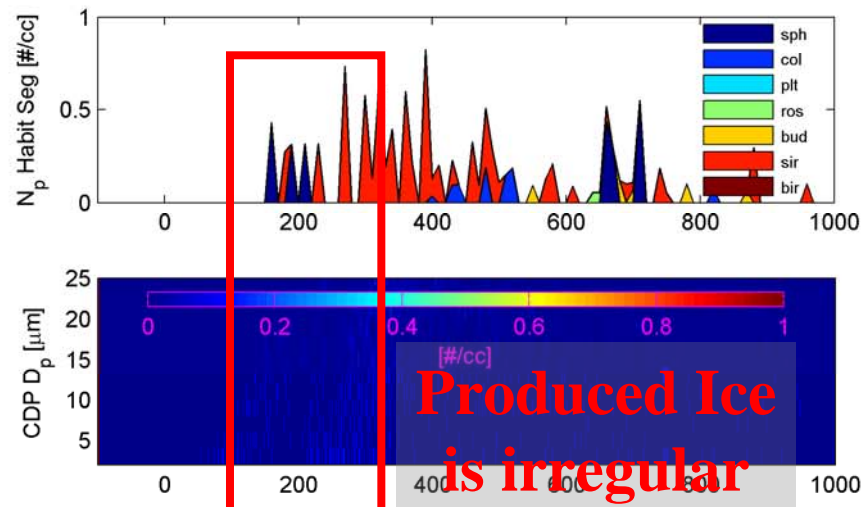
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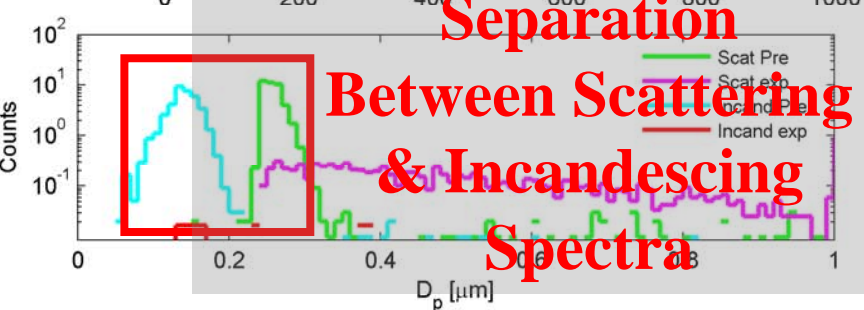
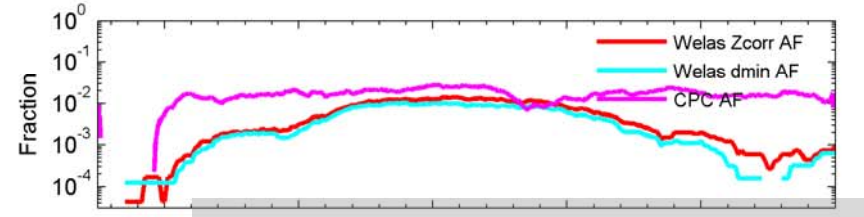
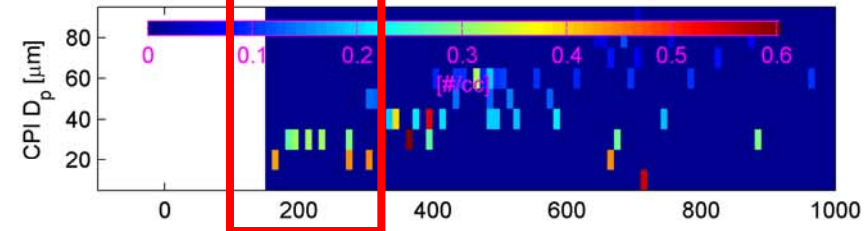
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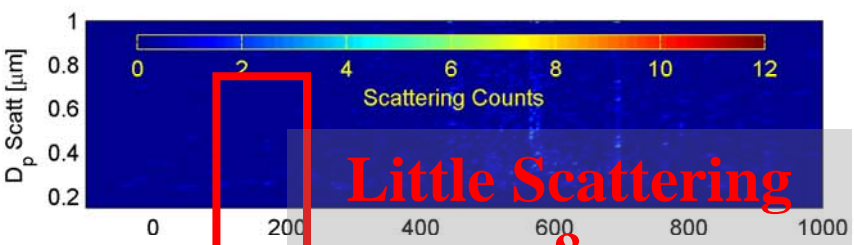
$S_{water} > 1$



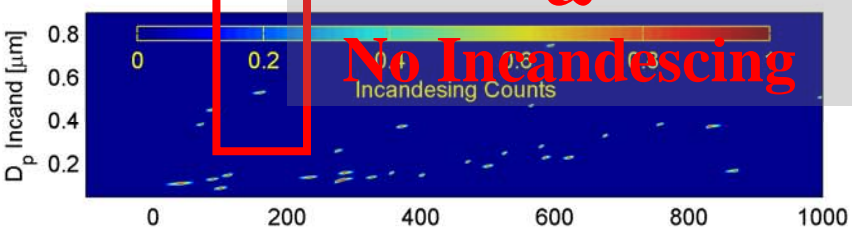
Produced Ice is irregular



~1% Activation



Little Scattering



No Incandescing

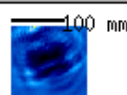
Separation

Between Scattering & Incandescing Spectra

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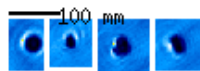
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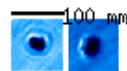
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20,00-40,00



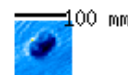
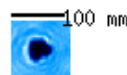
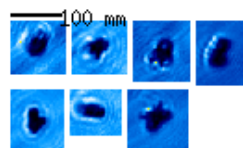
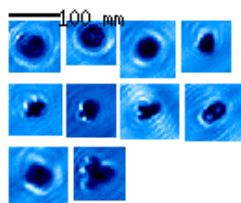
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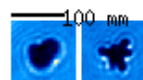
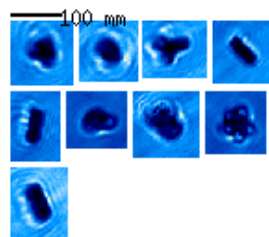
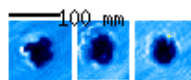
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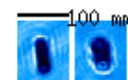


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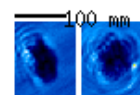
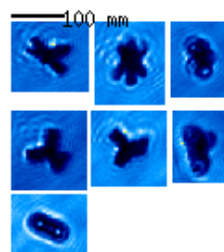
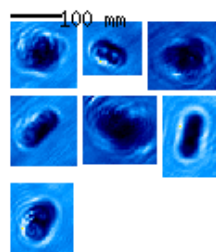


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80,00-100,00

No Data



No Data

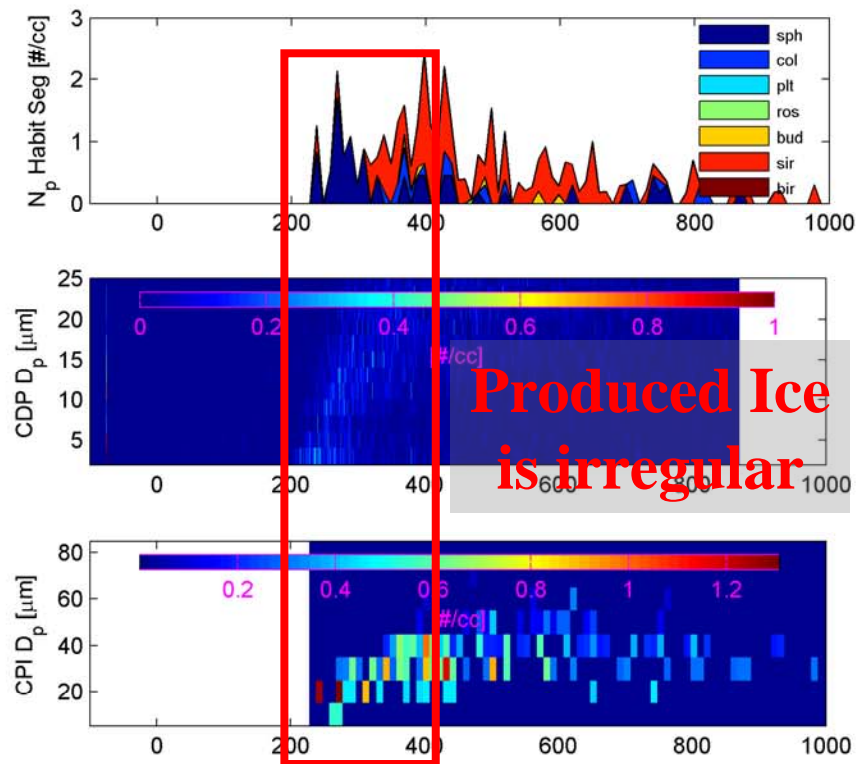
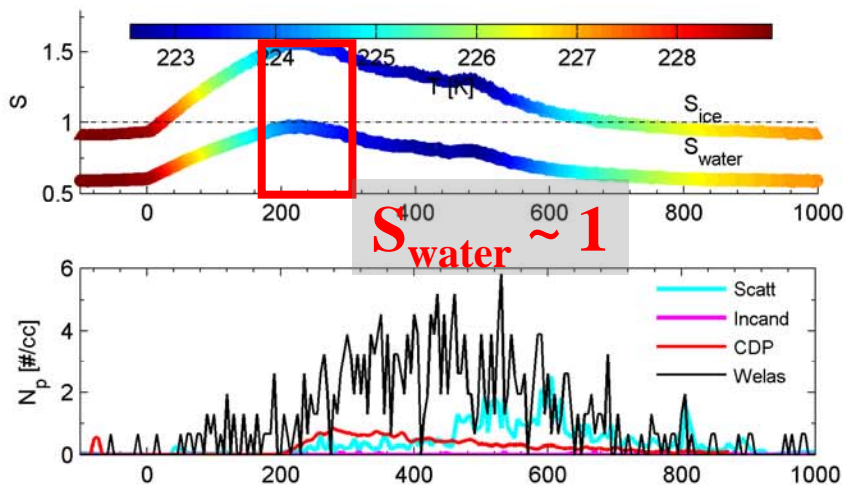
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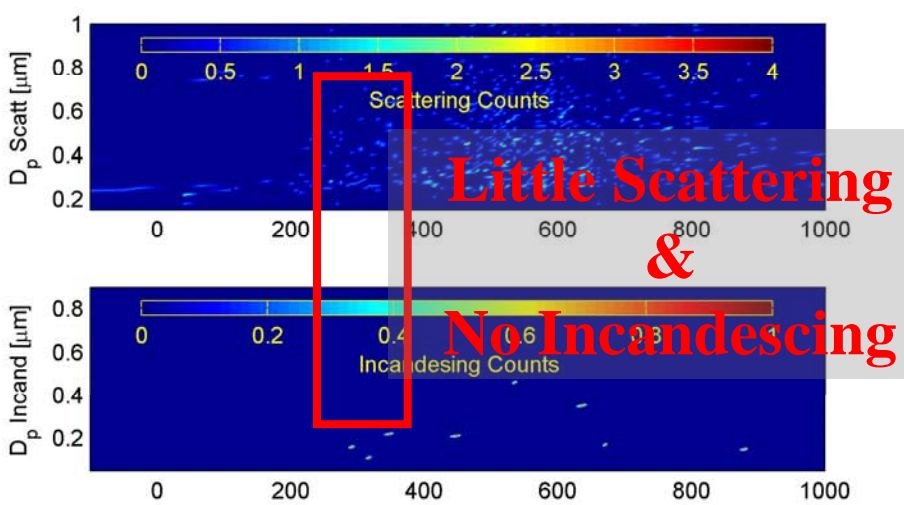
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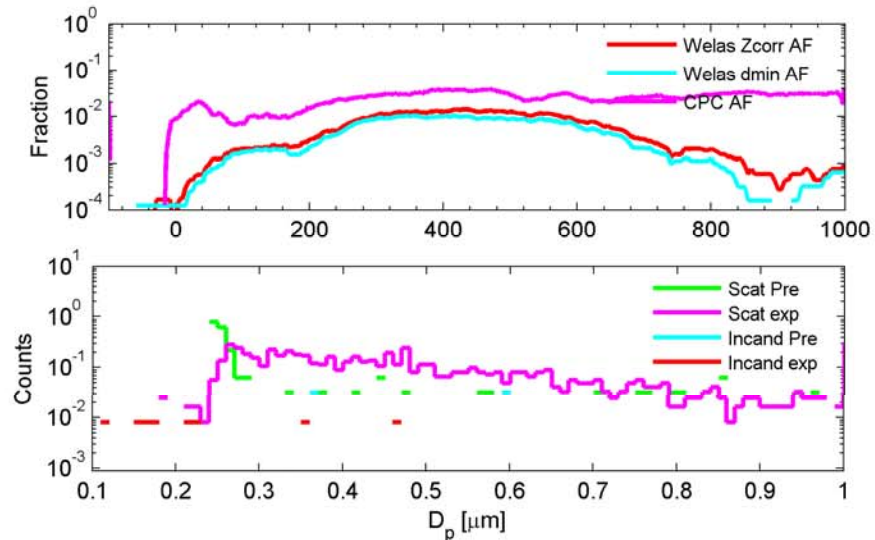



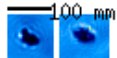

Produced Ice is irregular

~1% Activation

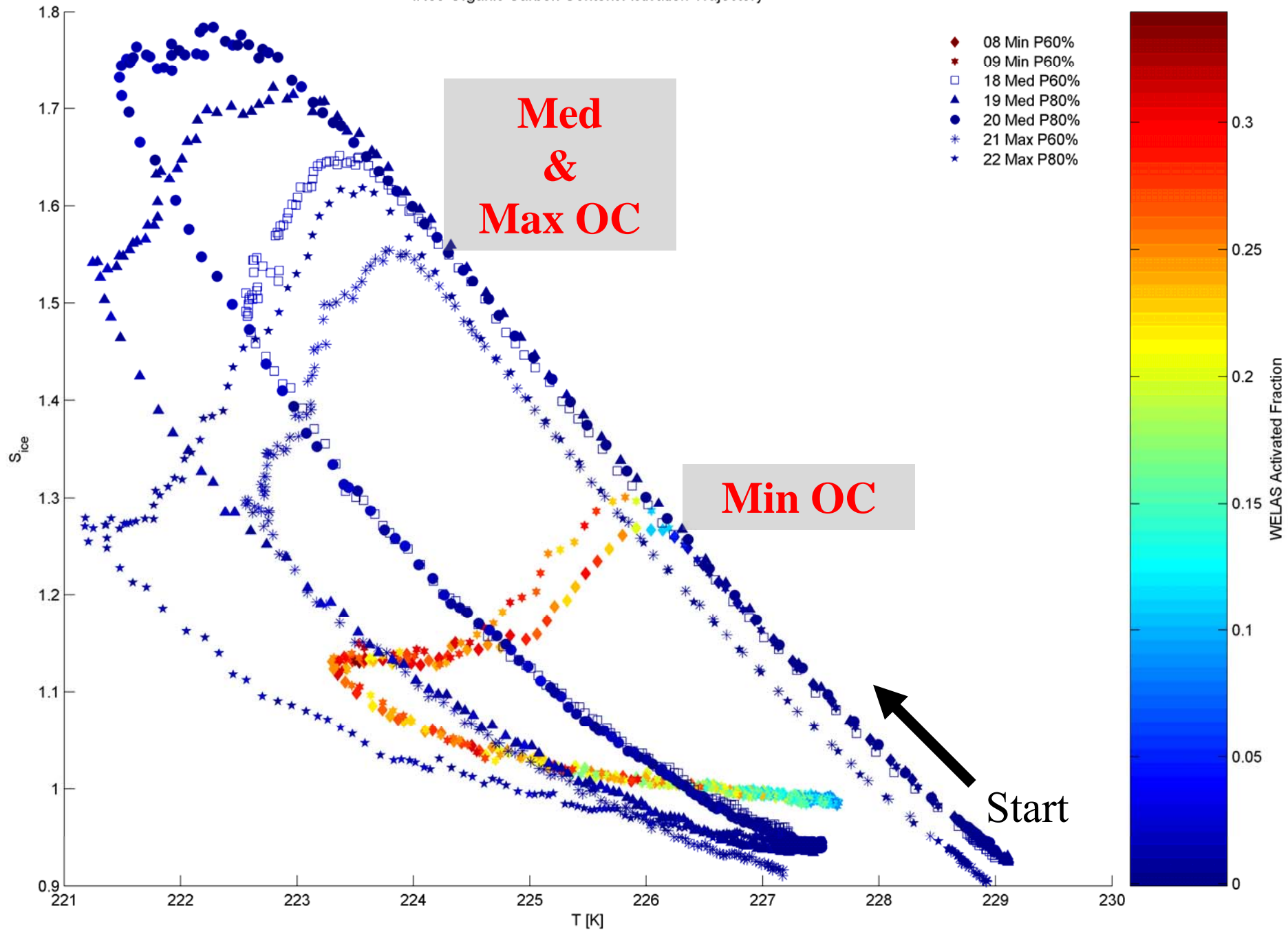


Little Scattering & No Incandescing



0,00-20,00	No Data	No Data	No Data	No Data	No Data	No Data
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40,00-60,00						
60,00-80,00						No Data
80,00-100,00	No Data				No Data	No Data
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IN09 Organic Carbon Content Activation Trajectory

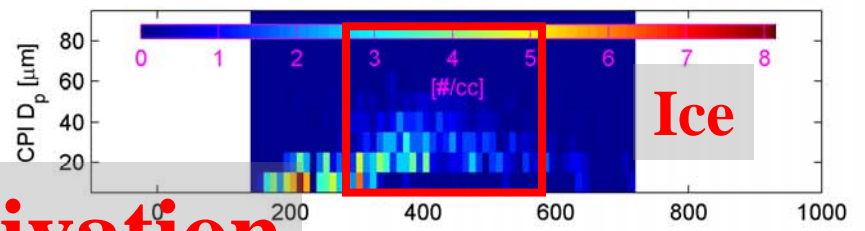
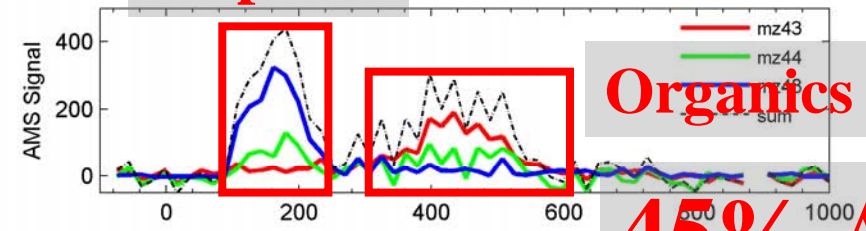
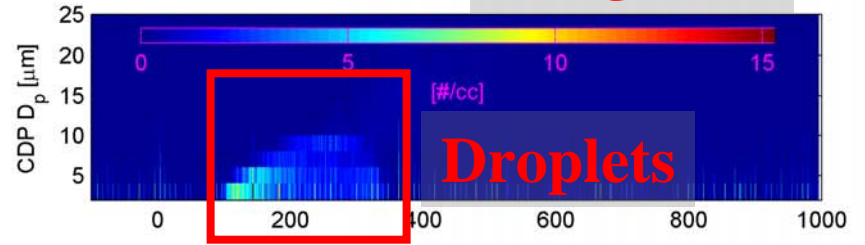
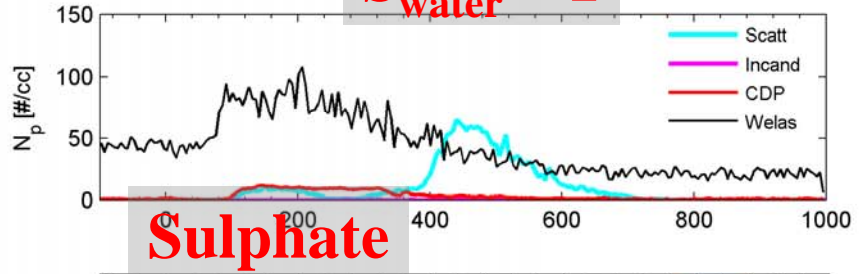
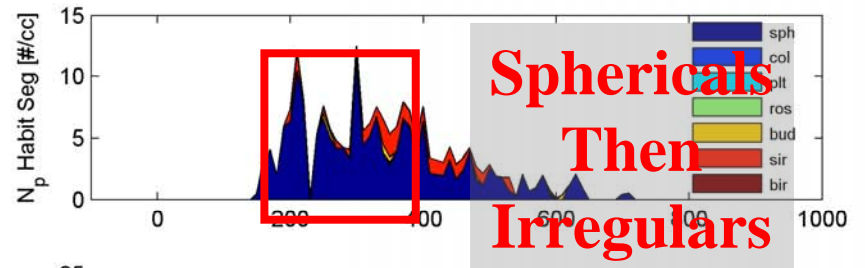
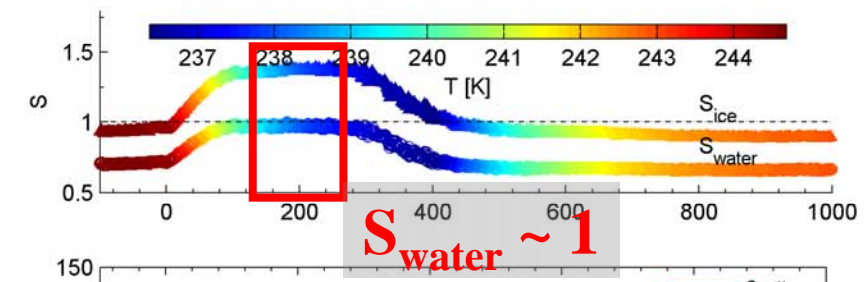


Expansion	Type	Pumping	Temperature	S _{ice} Peak	Temperature Peak	Activated Fraction
IN09_08	Min OC	60%	228K	1.30	226K	0.30-0.35
IN09_09	Min OC	60%	228K	1.30	226K	0.30-0.36
IN09_18	Med OC	60%	228K	1.60	223.5K	0.02-0.03
IN09_19	Med OC	80%	228K	1.80	222.5K	0.02-0.03
IN09_20	Med OC	80%	228K	1.80	222.5K	0.02-0.03
IN09_21	Max OC	60%	228K	1.55	224K	0.01
IN09_22	Max OC	80%	228K	1.62	223.5K	0.02-0.03

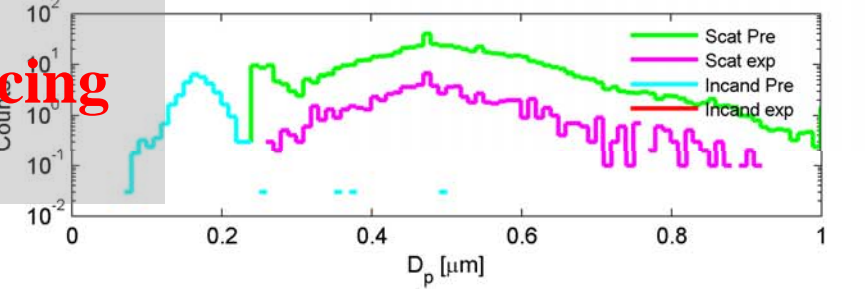
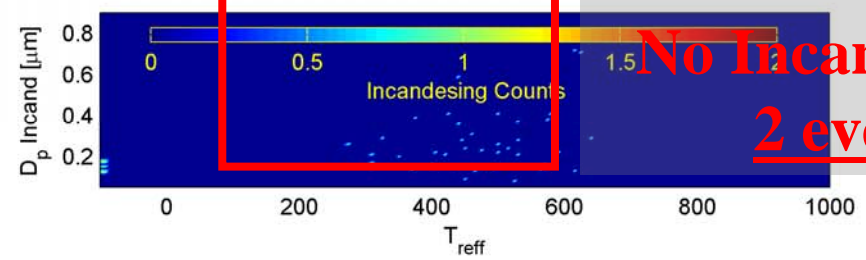
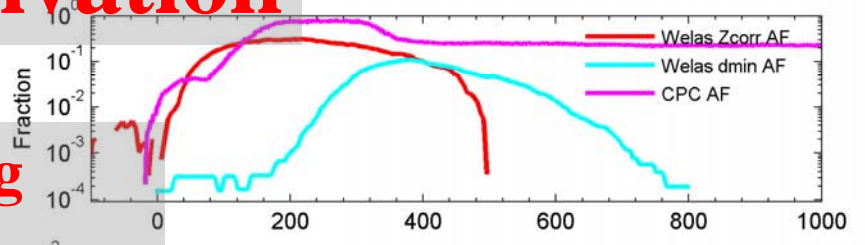
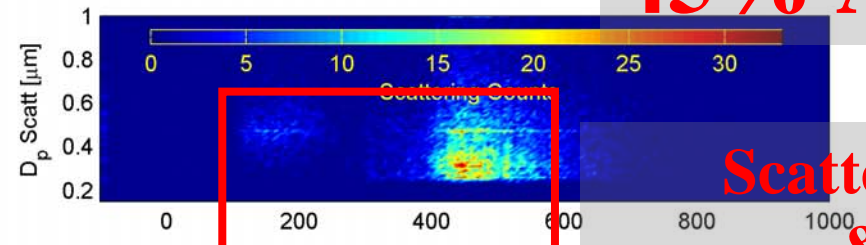
Min OC IN active – 30/35% AF, S_{ice} 1.30, 226K

**Med & Min OC NOT IN active – few% AF,
S_{ice} 1.80, 224K**

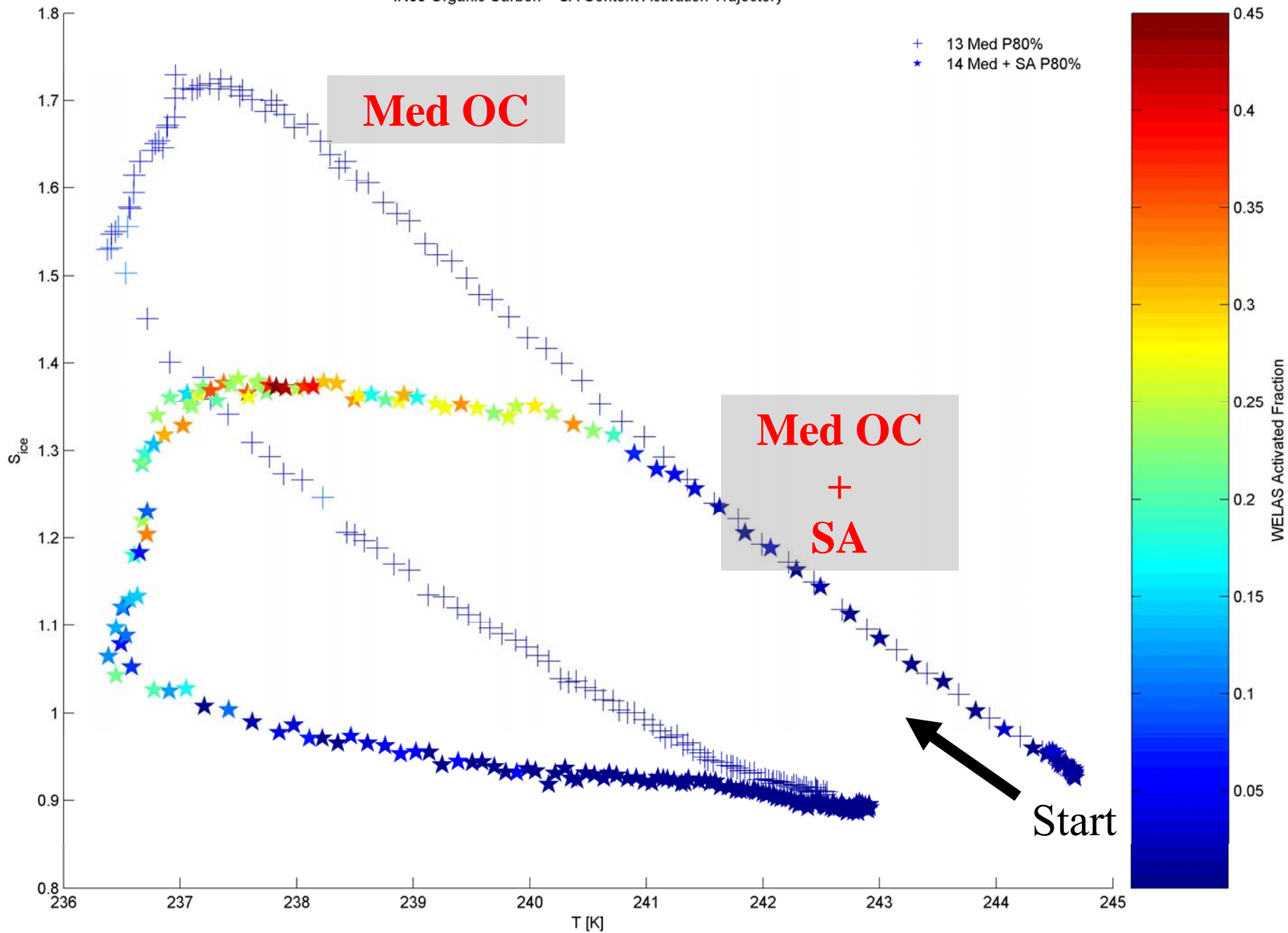
Influence of mixing with Sulphuric Acid



45% Activation



IN09 Organic Carbon + SA Content Activation Trajectory



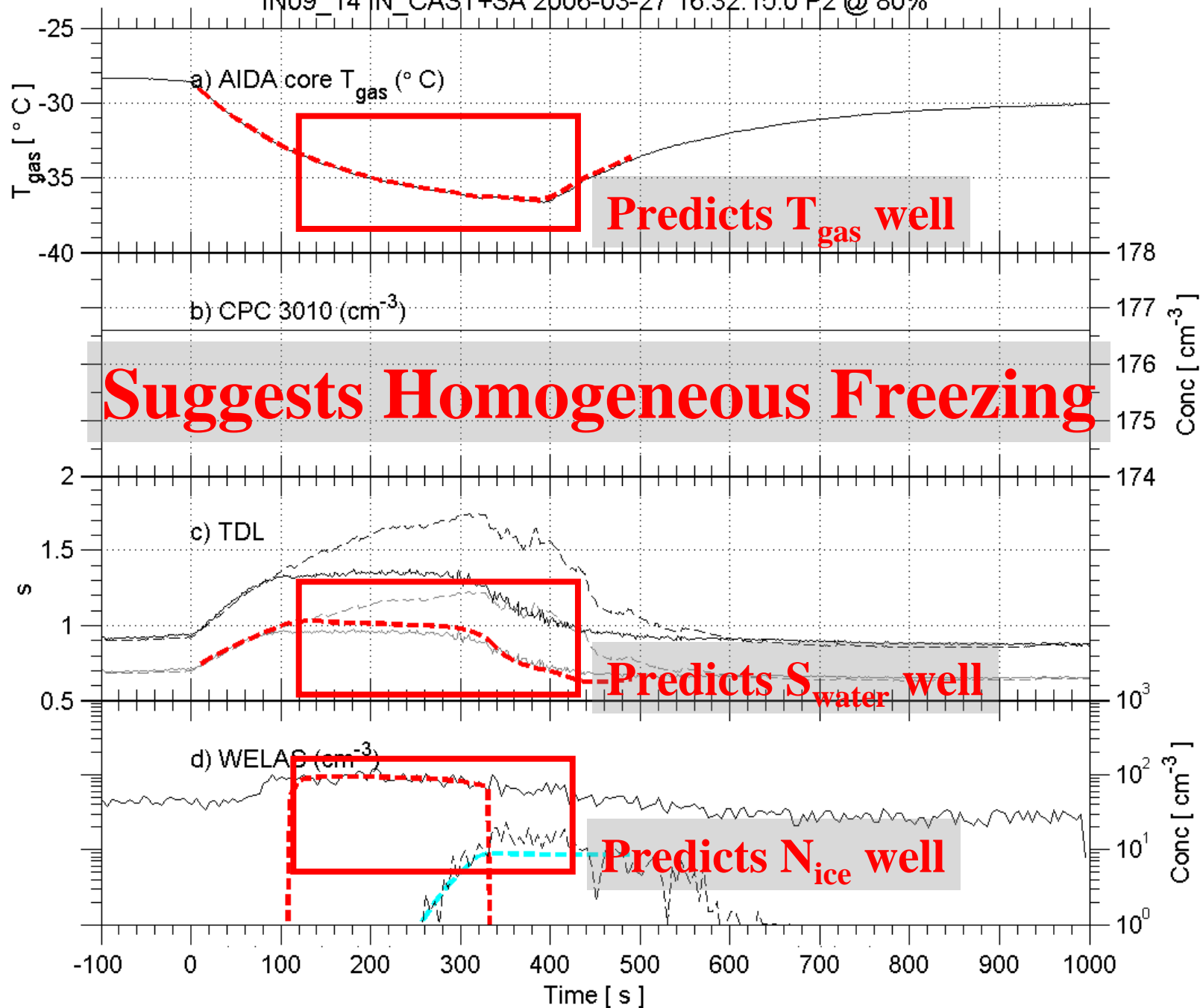
This ice could have come about by two obvious mechanisms:

1. The supercooled sulphuric acid droplets have homogeneously frozen. Soot had no part to play in the onset of freezing.

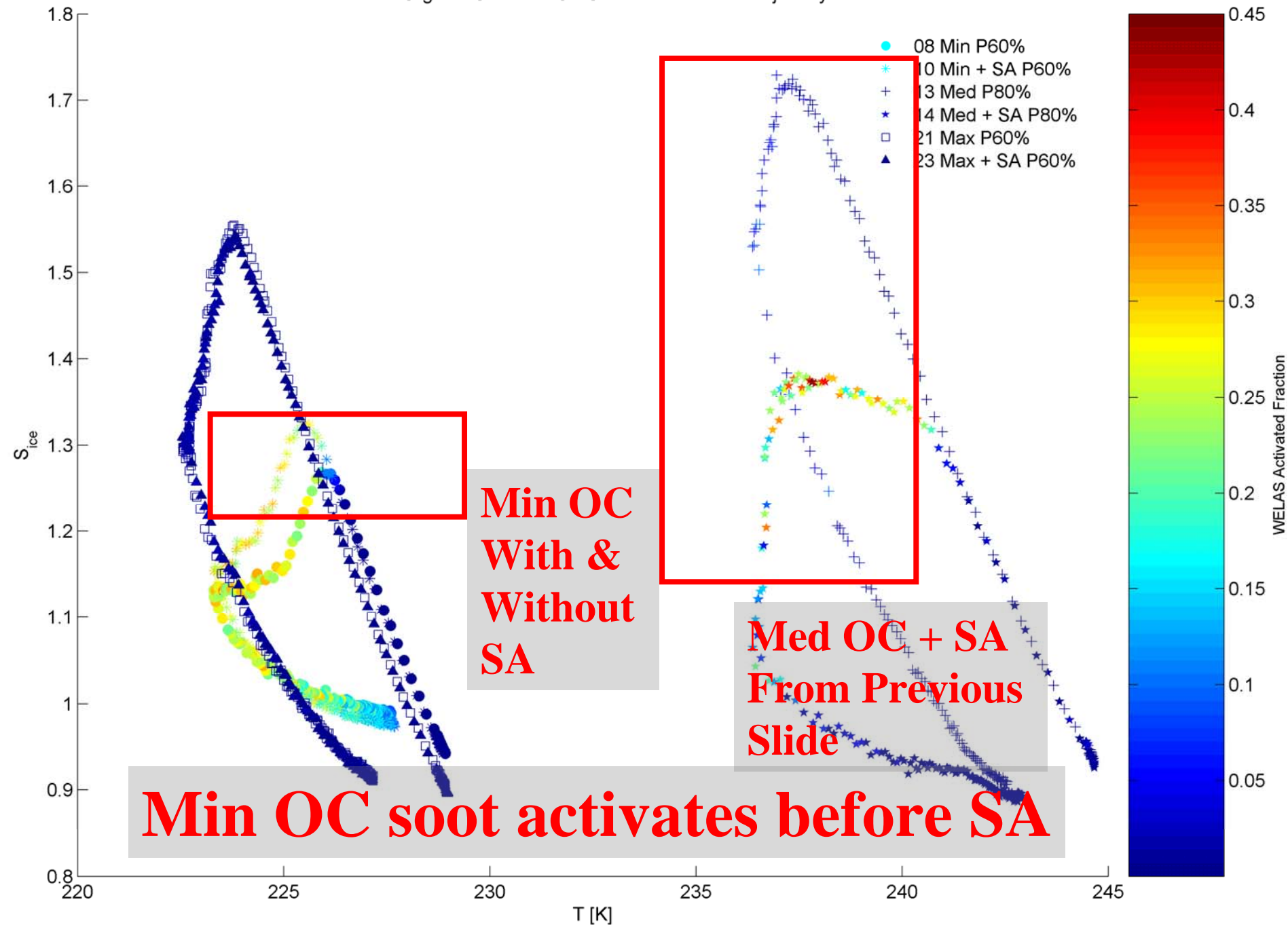
If this is true then it should be possible to model the observed freezing using the Koop parameterisation [Koop, et al., 2000].

2. Immersion and/or contact freezing mechanisms can occur under these conditions. Both soot and sulphuric acid are necessary.

DeMott has shown the flame soot from an acetylene burner to be an efficient immersion freezing nuclei once water had condensed upon it and that soot may catalyse heterogeneous freezing



IN09 Organic Carbon + SA Content Activation Trajectory



Conclusions

Only the minimum organic carbon content soot is significantly active as ice nuclei under untreated conditions and may be a significant source of IN.

Both the medium and maximum organic carbon content flame soot did not activate notably and can be considered to be poor ice nuclei.

Upon being externally mixed with sulphuric acid the minimum organic carbon content soot exhibited no change in activation properties, indicating that heterogeneous ice formation on soot of this type is preferable to the homogeneous freezing pathway.

