VI-ACI: Work Package L3 - Laboratory Experiments at ETH Zurich -CCN – IN – Hygroscopicity – Aerosol-MS

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Overview

<u>CCN – warm clouds:</u>

- Does size matter or chemistry?
- Hygroscopicity and chemical composition

IN – mixed phase and cold clouds:

- Which modes of IN are important?
- Chemical "features" of good IN
- Summary

CCN – warm clouds Atmospheric Effects : Milan

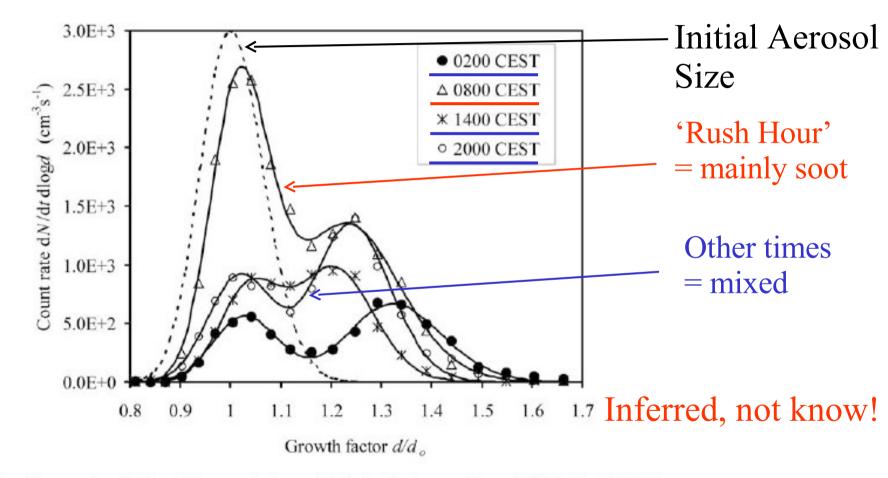
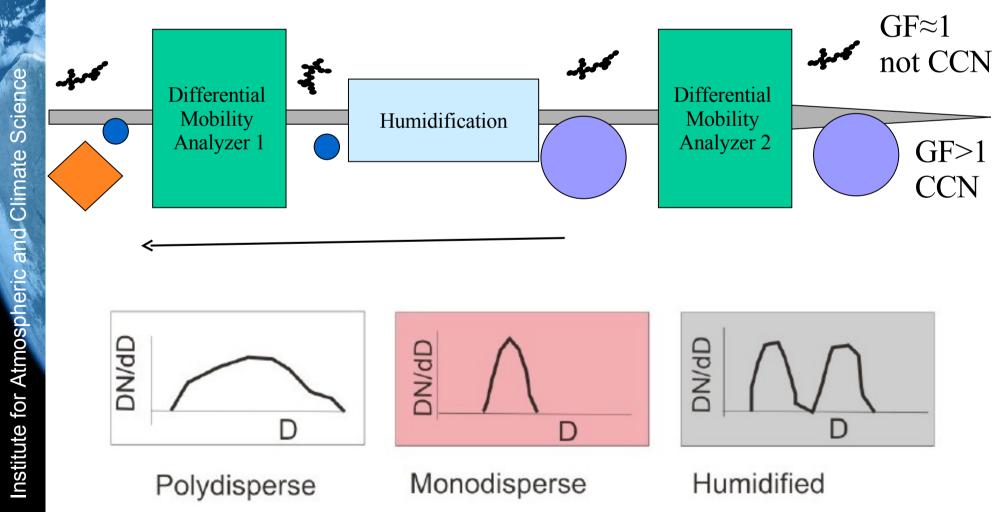


Figure 6. Temporal evolution of the growth factor (d/d_0) distribution on 2 June 1998 during IOP 2. Dry monodisperse particles ($d_0 = 100$ nm, dashed line) were exposed to RH = 90% at 0200, 0800, 1400, and 2000 CEST (points). The solid lines are bimodal fits of the less and more hygroscopic modes. *Y* axis units correspond to CPC counts normalized to both the diameter interval and the scan time during such an interval. Lines denote a monomodal/bimodal lognormal fit.

Baltensperger, JGR, 2002.

CCN – warm clouds

Experimental Setup: Hygroscopisity Tandem DMA (HTDMA)



CCN – warm clouds

Past Studies:

Examples of measurement results of GF (RH=85-90%)

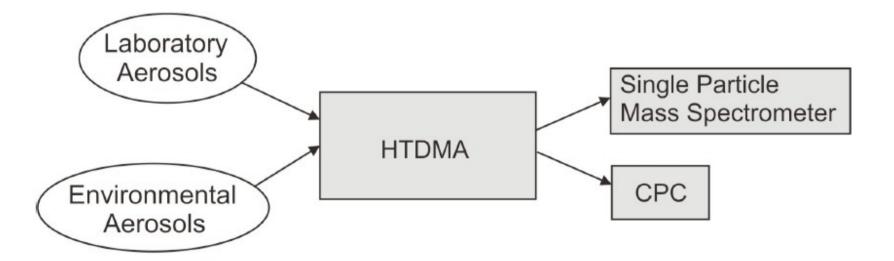
Location	GF	Group	
Milan	1.1 and 1.2-1.5	Baltensperger et al., 2002	
Jungfraujoch	1.1-1.3	Weingartner et al., 2002	
Izana, Tenerife	1.15, 1.40 and 1.61	Swietlicki et al, 2000	
Kleiner Feldberg	1.05 and 1.36	Svenningsson et al., 1994	
Munich	1.0 and 1.3	Ferron et al., 1994	

Determination of particle chemical composition:

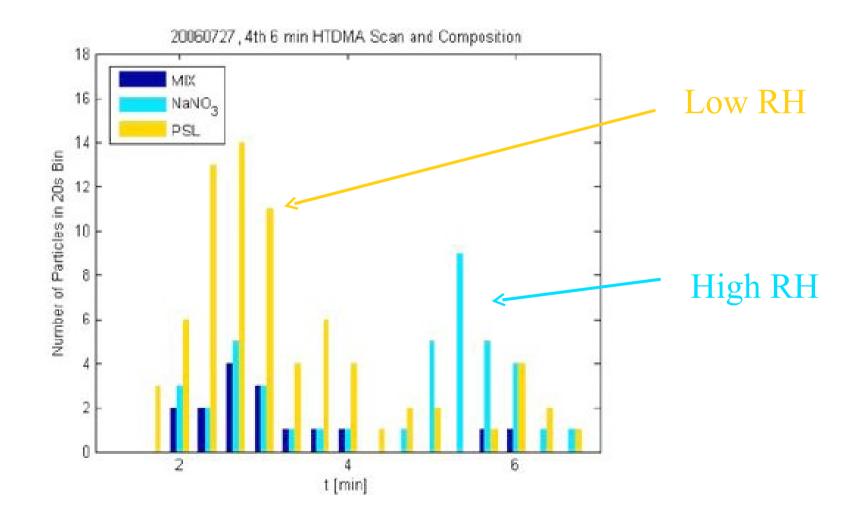
HTDMA and	Group	Limitation
Impactor	Pitchford & McMurry, 1994	parallel
Electron Microscopy	McMurry et al., 1996	parallel
SPLAT-MS	Buzorius et al., 2002	laboratory

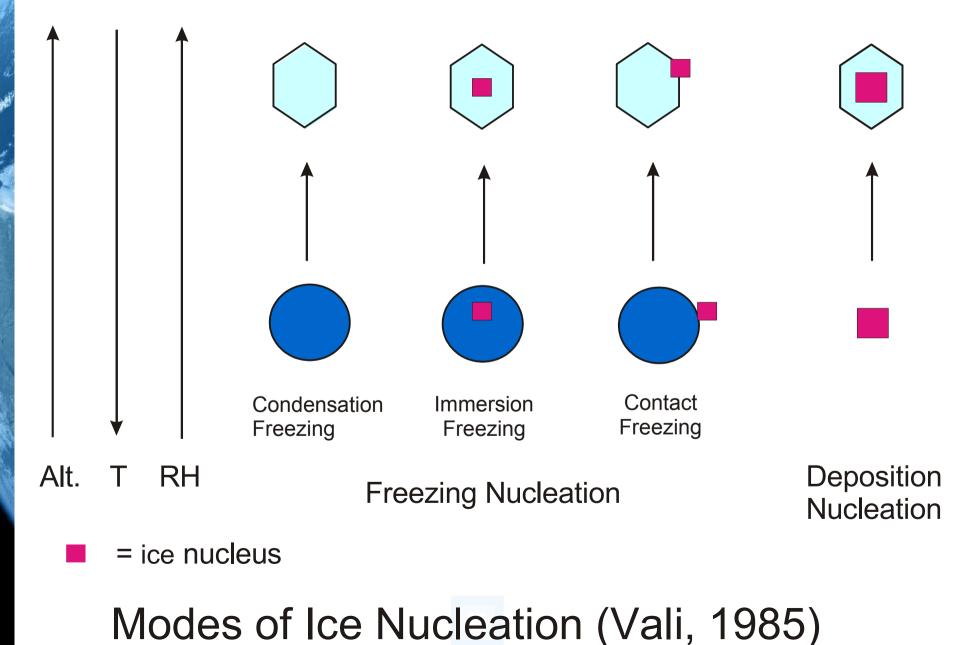
CCN – warm clouds

→ Coupling HTDMA with Single Particle-Mass-Spectrometry

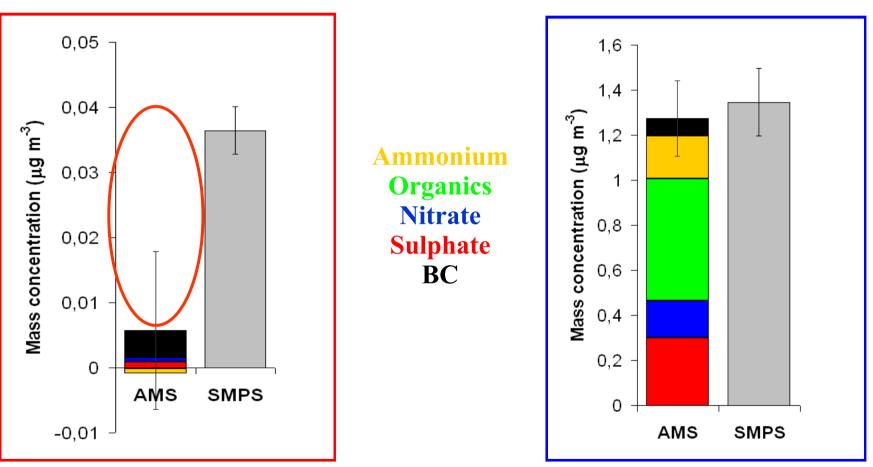


CCN – warm clouds HTDMA combined with Mass Spec: Proof of Concept?





What is the Aerosol and IN Composition ? Ice Residue Total

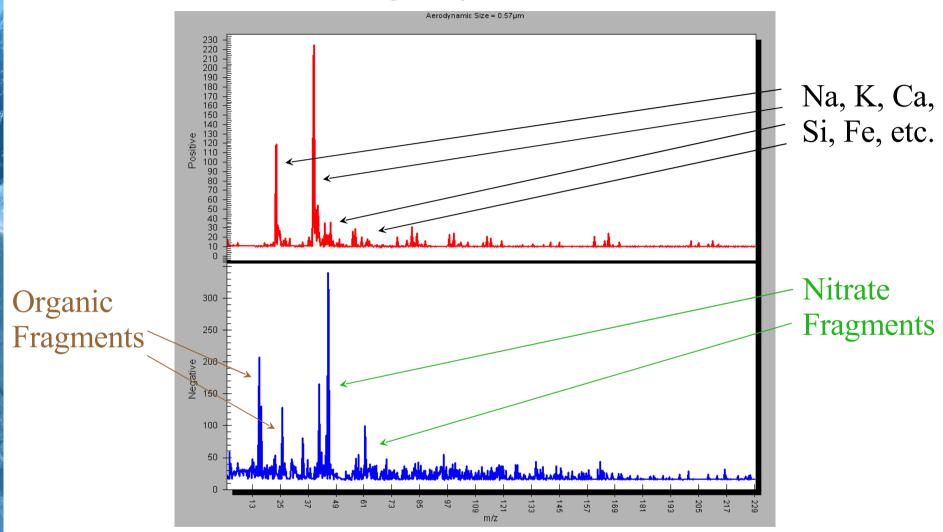


- Composite from AMS (volatile components) and BC (EC/soot) measurements.
- Lack of 'closure' between the volatile components + BC and ice residue.

What are the missing ice forming aerosols?

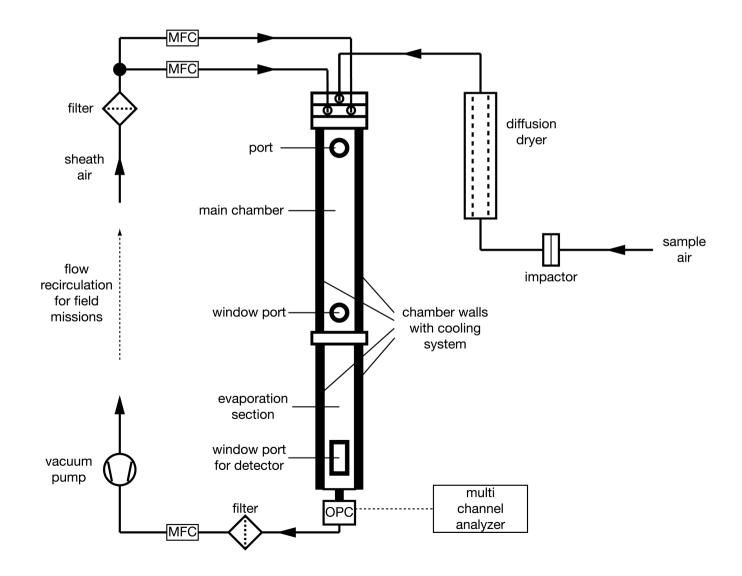
Courtesy of J. Schneider and S. Walter

2007 Ice Residuals – Jungfraujoch – measured with ATOFMS

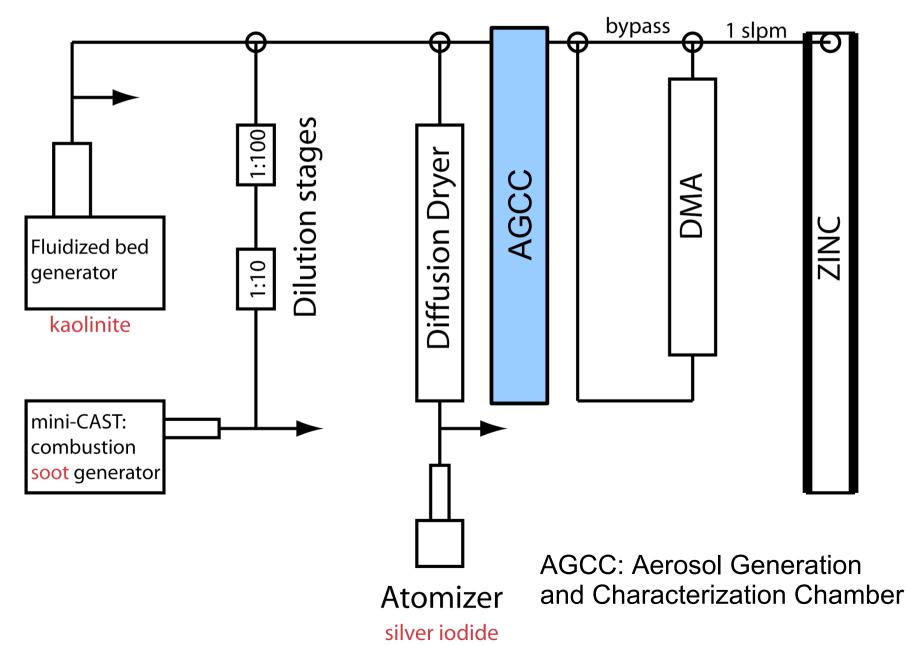


- Composition consistent with mineral dust which has uptaken some nitrate (see work of Grassian, Laskin, and others)
- No significant sulfate

The <u>Zurich Ice Nucleation Chamber (ZINC</u>) and its portable version PINC will be used for ice nucleation studies at ETH Zurich.

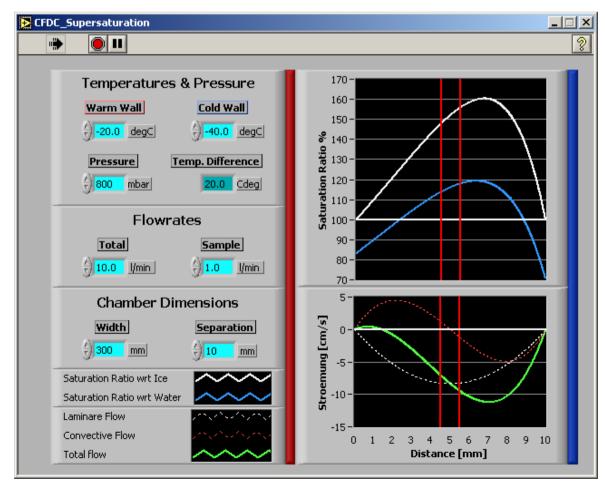


Setup for lab experiments with silver iodide, kaolinite and (soot)



Temperature and saturation ratio profiles in the ZINC/PINC instruments:

Both walls are held at different temperatures and are covered with ice. Linear profiles in absolute water vapour pressure and temperature develop. Because of the Clausius-Clapeyron-Law a supersaturation wrt ice exists in the chamber and peaks roughly at the same position where the aerosols are injected.



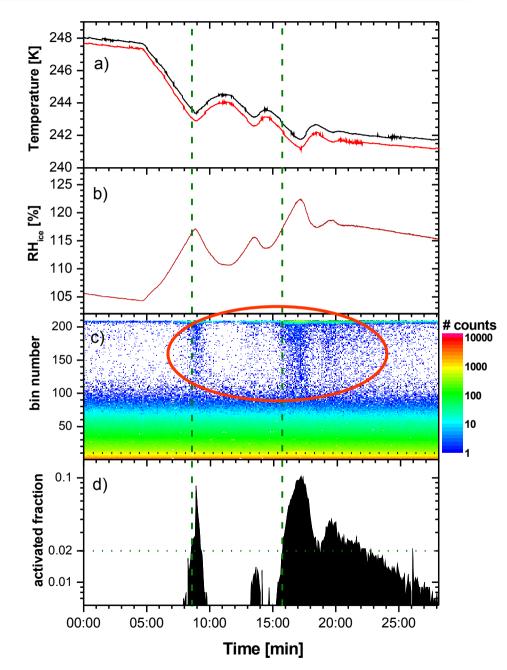
Proof of concept and validation experiments with Agl aerosols:

AgI particles were produced by atomizing an aqueous suspension.

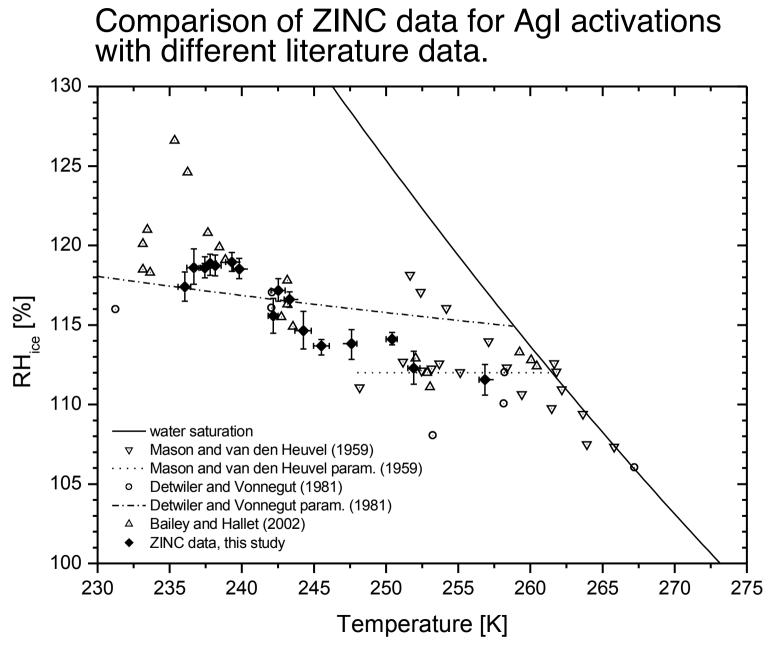
Temperatures of cold and warm walls were varied to scan through different sample temperatures (panel a) and supersaturations (panel b).

Particle size spectra were recorded (panel c) and integrated to get the fraction of activated particles (panel d) as function of T and RHi.

Activation threshold was set to 2 %.

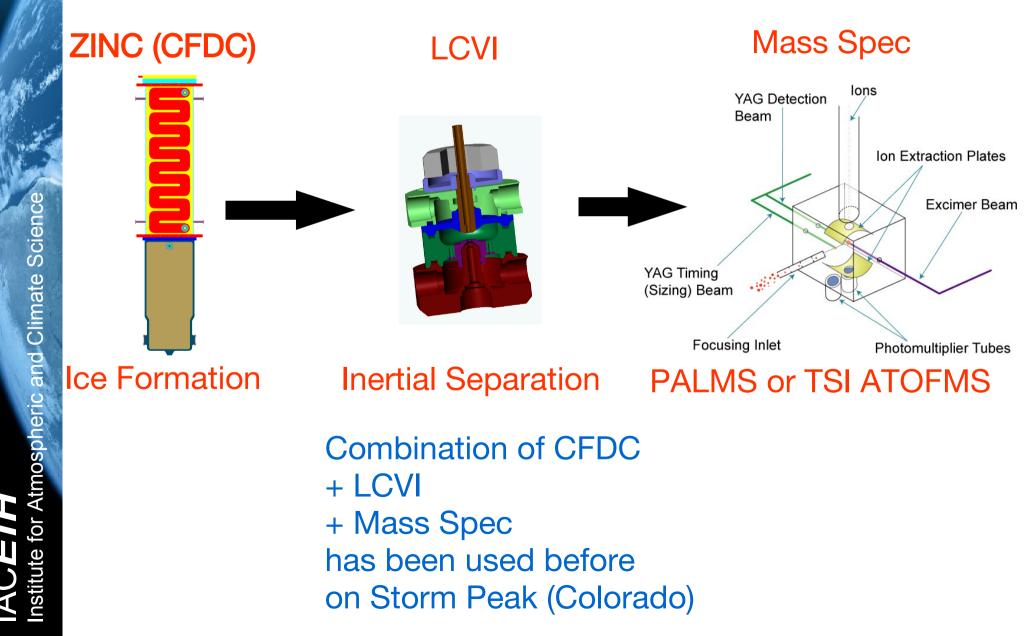


Stetzer et al., Aerosol Science & Technology 2007, submitted



Stetzer et al., Aerosol Science & Technology 2007, submitted

Mass Spectrometry of Ice Forming Aerosols



Summary

<u>CCN – warm clouds:</u>

- Aerosols with different growth factors (GF) have been observed in field studies
- A relationship between GF and chemical composition was only inferred but not directly measured
- A combination of HTDMA (PSI) and Aerosol Mass Spec (ETH) will be used for direct closure of GF and chemistry

IN – mixed phase and cold clouds:

- ZINC instrument for IN activation experiments available
- ZINC was tested and validated with AgI particles against literature data
- A LCVI will be used to separate activated ice crystals behind the ZINC instrument
- These particles will be directed to single particle aerosol MS to obtain chemical composition of individual IN